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1. What are the three  
different types of  
mixtures? 2. What is a  
solution? 3. Classify  
each of the following as  
a heterogeneous mixture  
or a homogeneous

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mixture. a. Salad \_\_\_\_\_

b. Tap water \_\_\_\_\_ c.

Muddy water \_\_\_\_\_ 4.

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and Colloids . It's all  
about the particles! •  
The size of the particles

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has a great effect on the properties of a mixture.

- You must observe a mixture's properties before deciding whether it is heterogeneous or homogeneous.

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Q.II. Classify each of  
the following mixtures

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as a solution,

suspension, emulsion or  
colloidal dispersion. 1.

A glass of whole milk .

2. A glass of pop ...

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PURE SUBSTANCES,  
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SUSPENSION, COLLOIDS**

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Mixtures: Solution or  
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worksheet is about  
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School subject: Science.

Grade/level:Grade 6.

Age: 11-12. Main

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A colloid is a

heterogeneous mixture

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in which the dispersed particles are intermediate in size between those of a solution and a suspension. The particles are spread evenly throughout the dispersion medium, which can be a solid, liquid, or gas.

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can be used in many

types of treatment

methods. When treating

a patient with an

ailment, this helps to

provide the type of

information a doctor can

use to guide the

treatment plan. These

are used in the diagnosis

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and treatment of many forms of illnesses.

Suspensions

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This quiz/worksheet combo will help test your understanding of the different types of mixtures, such as solutions, suspensions and colloids. Quiz questions ask you to

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identify the  
characteristics...

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Kaufmann. Ahead of

discussing Solutions

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remember to be aware

that Schooling can be

the step to a greater

another day, and also

learning does not only

quit as soon as the

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Solutions, Suspensions  
and Emulsions Liquid  
mixtures can be divided  
into 4 main types :

solutions, suspensions,  
colloids and emulsions

Solutions :-consist of  
soluble material or  
material (solute)



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dissolved in a liquid  
(solvent)-are clear-are  
homogenous (one  
phase) and do not settle

Worksheet

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A solution of 20% HCl  
and 80% water.

Chocolate milk Rubbing  
alcohol: 60% alcohol;

40% water. 1. Solution

2. Alloy 3. Dissolve 4.

Suspension 5. Colloid 6.

Insoluble A. When a  
substance cannot be dis-  
solved into a solution.

B. A mixture of two

metals. C. A mixture

that is homogeneous at

the molecular level. D.

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When something seems to disappear into a solution. E.

Suspensions

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